

## THE STUDY ON NANO TECHNOLOGY AND APPLICATIONS FOR SUPER CAPACITOR

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### **ABSTRACT**

*Electrochemical capacitors, otherwise called supercapacitors or ultracapacitors, have gotten a lot of consideration from innovative work to industrialization, attributable to their guarantee to convey undeniable degrees of electrical force and deal long working lifetimes. They are viewed as optimal possibility for energy stockpiling in high-power applications. Profiting from escalated nanoscale research in late many years, wonderful upgrades and advancement of supercapacitive energy stockpiling frameworks have been accomplished. Both the energy thickness and force thickness for supercapacitors have been considerably improved. In this audit article, we attempt to survey the significant effects of nanoscale research on the improvement of supercapacitors, as far as the generous improvement of capacitive execution for terminal materials, and progressive advances in anode and gadget setups. Likewise, late advancement in essential energy stockpiling instruments and models of supercapacitors are additionally checked on, including a new dynamically preferred intercalation component presented interestingly. The audit finishes up with depictions of the exhibition of as of now acknowledged viable uses of industrially accessible supercapacitor gadgets, particularly zeroing in on genuine utilization in vehicles that are exceptionally expected by future networks to additionally increase the wide consideration on clean energy stockpiling frameworks.*

**Keywords:** Nanotechnology, Super capacitor

## INTRODUCTION

The CNT/leading polymer composite arranged by polymerization of pyrrole on the nanotubes with ammonium persulfate as an oxidant displayed capacitance esteem up to 180 F/g, 112 while SWNT/polypyrrole composite showed capacitance esteem up to 260 F/g.<sup>113</sup> In two separate examinations, it was shown that the capacitance of an unoriented MWNT/polypyrrole composite was at any rate multiple times higher than that of one or the other part. Supercapacitor conduct of adjusted varieties of CNTs covered with polypyrrole has additionally been accounted for. Adversely charged MWNTs/polypyrrole layers were additionally explored by co-testimony through electrochemical oxidation of pyrrole. Electrochemical quartz gem microbalance (EQCM) affirmed the redox pseudocapacitance reaction of the polypyrrole with a system including unbending intercalation/deintercalation of counter particles into the polymer layer. The capacitance of the layer composite was accounted for to be in excess of 200 F/g in potassium chloride and tetrabutylammonium bromide arrangements. CNT networks can give a mesoporous platform, on which a permeable layer of polypyrrole can be electrodeposited; henceforth high charge elements during supercapacitor execution can be accomplished. A MWNT/poly (3-methylthylthiophene) composite based supercapacitor was assessed in 1 M LiClO<sub>4</sub> acetonitrile arrangement, and its particular capacitance was around 80 F/g. A MWNT/poly(3,4-ethylenedioxythiophene) composite, arranged by compound or electrochemical polymerization, when assessed in 1 M H<sub>2</sub>SO<sub>4</sub>, 6 M KOH, or 1 M tetraethylammonium tetrafluoroborate (TEABF<sub>4</sub>) in acetonitrile, displayed great cycling execution, and moderate capacitance esteems in the reach from 60 to 160 F/g. The capacitance of SWNT/polyaniline (PANI) composite manufactured by in-situ electrochemical polymerization was 310 F/g, higher than that of unadulterated PANI, on the grounds that the intricate construction in this composite offered more dynamic locales for Faradaic responses.

Among the carbon nanotube based supercapacitor terminals, as of late revealed approaches utilizing high temperature carbonization or potentially CO<sub>2</sub> actual initiation of carbon nanotube/polymer paired composite forerunners show extremely encourage outcomes. The pellet anodes produced using carbonized SWNT/poly(vinylidene chloride) (7:3) composites assessed in 7.5 M KOH electrolytes give estimations of 180 F/g explicit capacitance, 20 kW/kg power thickness and 7 Wh/kg energy thickness individually. SWNT/polyacrylonitrile

(60:40) composite film gives a one of a kind handle on the pore design and surface territory to bring about elite.

## OBJECTIVE OF THE STUDY

1. To examine CNT application in supercapacitors.
2. To acknowledge pore structure control of CNT based supercapacitor terminals.

## CARBON NANOTUBE

Carbon nanotubes are extremely alluring both for mechanical and scholastic examination because of their extraordinary properties, as has been audited by Baughman et al. furthermore, Popov. In 1991, Iijima originally noticed the multi-walled carbon nanotubes (MWNT). In 1993, Iijima and Ichihashi and Bethune et al. detailed the perception of single divider carbon nanotube (SWNT) nearly simultaneously. Union procedures of carbon nanotube incorporate circular segment release, laser-removal, 83 and compound fume disintegration (synergist development), and High-pressure CO disproportionation (HiPco). The mechanical properties, electronic conductivity, warm conductivity and optical properties of carbon nanotubes are extremely great. The rigidity and Young's modulus of individual SWNT can be just about as high as 37 GPa and 640 GPa. Electrical conductivity of SWNT is of the request for 10<sup>6</sup> S/m at room temperature, and 5000 S/m for adjusted MWNT film along the cylinder pivot. Some SWNTs with little distance across additionally display superconductivity at low temperature, for example, 1.4-nm-width SWNT with change temperature at 0.55 K, and 0.5-nm-measurement SWNT developed in zeolites with progress temperature at 5 K.

Warm conductivity of individual MWNT (> 3000 W/m.K)<sup>78</sup> is higher than that of the common jewel and that of the basal plane of graphite (2000 W/m.K), however lower than that of the separated SWNT (6600 W/m.K) at room temperature, which is equivalent to the warm conductivity of a speculative detached graphene monolayer. Optical properties of SWNT have been dissected in detail, and the mean breadth and width dissemination in mass SWNT tests can be precisely decided from a joined investigation of optical assimilation, high-goal electron energy-misfortune spectroscopy in transmission, and tight-restricting computations.

The particular surface territories of individual carbon nanotubes and groups of carbon nanotubes have been determined as an element of cylinder measurement, the quantity of dividers and the quantity of carbon nanotubes in a pack. The particular surface region of an individual cylinder with 1 nm distance across can be just about as high as 1315 m<sup>2</sup>/g, while 400 m<sup>2</sup>/g for the groups of 7 nm measurement (for SWNT of 1 nm width).

In light of these properties, various applications or potential applications have been accounted for carbon nanotubes, like conductive and high-strength composites, hydrogen stockpiling media, energy stockpiling and energy transformation gadgets, field outflow showcases, and radiation sources, and nanometered-sized semiconductor gadgets, sensors, tests, interconnects, field-impact semiconductors, single electron semiconductors, amending diodes.

### **CARBON NANOTUBE BASED SUPERCAPACITOR**

The announced explicit capacitance estimations of the carbon nanotubes range from 20 F/g to > 300 F/g. The supercapacitor cathodes arranged from MWNT were first revealed by Niu et al.<sup>35</sup> with explicit capacitance, power thickness and energy thickness of estimations of 113 F/g, 8 kW/kg and 0.56 Wh/kg, separately, when 38 wt% H<sub>2</sub>SO<sub>4</sub> watery arrangement was utilized as electrolyte. Mama et al.<sup>36</sup> likewise detailed MWNT supercapacitor cathodes, which have generally low explicit capacitance estimations of 15-25 F/g additionally in 38 wt% H<sub>2</sub>SO<sub>4</sub> arrangement. The SWNT terminals as bucky paper give a particular capacitance estimation of 40 F/g in 6 M KOH fluid solution, 29 and 20-40 F/g in NaCl watery electrolyte. Similar outcomes were additionally detailed in 1.0 M LiClO<sub>4</sub>/propylene carbonate electrolytes by Shiraishi et al.<sup>100</sup> However, Liu et al.<sup>101</sup> revealed that the particular capacitance of SWNTs in acetonitrile electrolyte containing 0.1 M tetra-n-butylammonium hexafluorophosphate is around 280 F/g.

Physical and substance enactment are general practices for creating huge explicit surface zone initiated carbon (~2000 m<sup>2</sup>/g). This methodology was likewise utilized for the treatment of carbon nanotube based terminals to improve the particular surface zone by acquainting micropores with upgrade the capacitance execution. Truly enacted and artificially actuated carbon nanotube anodes were accounted for to have altogether higher explicit surface zone

just as the particular capacitance over non-initiated terminals. Other than pore constructions and explicit surface territory, the wetting ability of the terminal materials is likewise a significant property that influences the capacitance conduct. Thus, hot nitric corrosive fluorine, or alkali plasma functionalized carbon nanotube anodes have been tried for capacitance execution. Be that as it may, the pseudofaradaic responses instigated by the surface utilitarian gatherings lead to insecure capacitance and expanded current spillage.

Leading polymers, for example, polyacetylene, polypyrrole, polyaniline, polythiophene, and their subsidiaries are additionally basic cathode materials for supercapacitors. The alteration of CNTs with directing polymers is one approach to build the capacitance of the composite coming about because of redox commitment of the leading polymers. In the CNT/leading polymer composite, CNTs are electron acceptors, while the directing polymer fills in as an electron benefactor. A charge-move complex is shaped between CNTs in their ground state and aniline monomer. Various CNT/directing polymer composite examinations for electrochemical capacitor application have been accounted for.

## **CARBON NANOMATERIALS COMPOSITES**

The utilization of carbon nanostructures composites is likewise examined as terminal material by expanding the conductivity and available surface territory. Nonetheless, the composites of CNTs, CNFs, and graphene are engaged here.

The CNTs are the materials with exceptional electrical properties and can be utilized to make composite with improved electrical conductivity. The high conductivity helped simple charge move among pores and surface. The composites of CNTs and PAN nanofibers were made by electrospinning CNTs with PAN followed via carbonization and initiation in hydroperoxide at 700°C. The conductivity of the composite ( $5.32 \text{ S cm}^{-1}$ ) was higher than just PAN-CNFs ( $0.86 \text{ S cm}^{-1}$ ). The particular surface territory and explicit micropores volume of composite ( $810 \text{ m}^2/\text{g}$ ) ( $0.135 \text{ cm}^3/\text{g}$ ) were lower than PAN-CNFs ( $930 \text{ m}^2/\text{g}$ ) ( $0.230 \text{ cm}^3/\text{g}$ ) however the mesopores volume of the composite was higher ( $0.159 \text{ cm}^3/\text{g}$ ) than PAN-CNFs ( $0.146 \text{ cm}^3/\text{g}$ ). Higher open SSA of the composite brought about higher explicit capacitance ( $310 \text{ F/g}$ ) than PAN-CNFs ( $169 \text{ F/g}$ ).

The high mesopores volume likewise gave high rate ability (90%) when release current thickness was expanded from 100 to 1000 mA/g. Also, PAN-CNFs and SWCNTs composite was made in comparative manners yet absorbed HNO<sub>3</sub> to eliminate metal particles. The SSA got from the composite was lower (132 m<sup>2</sup>/g) however with higher conductivity (8.82 S cm<sup>-1</sup>) because of the CNTs. The particular capacitance from the composite was high (417 F/g). Notwithstanding inserting by electrospinning, the CNTs are likewise developed straightforwardly on carbon nanofibers by CVD with impetus molecule implanted by electrospinning. Like carbon nanotubes, the carbon nanofibers were additionally used to make composite. The conductivity and SSA of the composite expanded when CNFs were utilized with permeable nanosheet from bacterial cellulose. The conductivity of composite expanded to 10.1 S m<sup>-1</sup> contrasted with nanosheet of 7.6 S m<sup>-1</sup>.

## **CHANGE METAL OXIDES AND SULFIDES**

Different change metal oxides with various oxidation states address such a charming materials for supercapacitors inferable from the remarkable helper flexibility and high express capacitance. For the advancement metal oxides, energy stockpiling relies upon the reversible redox reaction despite the electric twofold layer stockpiling. Thusly, pseudocapitance is the common part in control stockpiling measure. Ruthenium oxides, manganese oxides, vanadium pentoxide, nickel oxides, cobalt oxides et al. have been investigated as cathode materials for supercapacitor applications

The most powerful construction is hydrous ruthenium oxide in light of its natural reversibility for various surface redox couples and high conductivity. Ruthenium oxide terminals with unquestionably amassed structure, glasslike stage, atom size and morphology have been seemed to show extraordinary capacitive properties (Simon and Gogotsi, 2018). The specific capacitances for hydrous ruthenium oxides in the extent of 200 to 1300 Fg<sup>-1</sup> have been represented (Chang et al., 2009). Past examinations showed that the electrical and capacitive properties of ruthenium oxides were associated with the proportion of water in the materials and the degree of crystallinity (Fu et al., 2002). A movement of ruthenium oxide nanoparticles with different water substance were analyzed (Yuan et al., 2009). Results displayed the meaning of hydrous areas (either interparticle or interlayer) to allow extensive protonic conduction for high energy and high force supercapacitors. Lately, mesoporous

RuO<sub>2</sub> meager movies were found to show a specific capacitance of around 1000 F/g at a yield speed of 10 mV/s. The high capacitance may have started from the characteristic thought of hydrous ruthenium oxide and the high mesoporosity (Capucine et al., 2009). The critical disadvantage of ruthenium oxide as supercapacitor cathode is the high cost on account of the obliged openness of ruthenium.

## TRANSITION METAL SULFIDES

Like metal oxides, metal sulfides (MSs) are moreover electrochemically powerful terminal materials for supercapacitors. Over the earlier decade, nanostructured metal sulfides have been comprehensively investigated as anode materials for EC structure considering the way that the extraordinary exhibition game by redox reactions of cathode commonly occur in metal sulfides where metal particles have diverse valence states. CuS, MoS<sub>2</sub>, Bi<sub>2</sub>S<sub>3</sub> and NiS nanoparticles are extensively used in supercapacitors.

Among the diverse metal sulfides, cobalt sulfide has been read for its relative straightforwardness, extraordinary capacitance, low toxic substance levels and colossal surface measured than others. The gathering of cobalt sulfide, for instance, CoS, CoS<sub>2</sub>, Co<sub>3</sub>S<sub>4</sub> and Co<sub>9</sub>S<sub>8</sub> has been read for supercapacitor anode materials. Lou et al., have point by point CoS<sub>2</sub> ellipsoids, which showed captivating supercapacitive properties of 1040 F/g. Another assessment pack uncovered a most extraordinary unequivocal capacitance of 508 F/g (Yang et al., 2011). Xu et al considered the advancement of Co<sub>9</sub>S<sub>8</sub> bunches and the examples showed most extraordinary express capacitance of 113.F/g (Xu et al., 2013).

## Capacitance Evaluation

The capacitance execution, regular voltage - time reaction in CC estimation, is appeared in Figure 2.8. Comparable voltage subordinate capacitance conduct was additionally seen in the carbonized and actuated SWNT/PAN film terminals. The voltage-subordinate explicit capacitance of PAN/SWNT/SAN composite movies is accepted to identify with their expansive pore size appropriation. Likewise, the voltage reliance of the electrical twofold layer limit and the pseudo-capacitance, which are the significant charge stockpiling components for carbon-based supercapacitor anodes, probably won't be ignored by the same token. The cyclic voltammetric bends of all carbonized PAN/SWNT/SAN film display

twisted rectangular shape, which might be ascribed to the presence of useful gatherings or potentially to expansive pore size dissemination.

### **Capacitance Performance**

The plots of explicit capacitance versus releasing current thickness for different examples at 0.1 V are given in Figure 2.11. At comparative releasing current, CC and CV estimation give tantamount explicit capacitance esteems. For all the examples, the particular capacitance monotonically diminished with expanding releasing current thickness. Notwithstanding, the declining pace of the particular capacitance at little releasing current shifted from one example to another. The declining rate for PAN3-SAN30, PAN1-SAN50, and PAN1-SAN10, which have generally enormous explicit capacitance at high releasing current, is essentially lower than that for PAN1-SAN30 and PAN2-SAN30. Broad examinations on the capacitance execution of permeable cathodes uncover that, because of the covering impact of the electrical twofold layer, electrolyte admittance to the more modest pores is restricted. A cut-off pore size of up to a few nanometers, contingent upon the electrolyte focus and the applied voltage, has been anticipated as examined in Section 1.6. Just those pores bigger than the cut-off pore size can be gotten to by the electrolyte to frame the electrical twofold layer. Further, the higher obstruction of the electrolytes in more modest pores

### **GRAPHENE NANOSHEETS FOR SUPERCAPACITOR APPLICATIONS**

Graphene is a single atom thick two dimensional nano structured material contains  $sp^2$  carbon atoms. It has unique properties such as large surface area, good electronic conductivity, high thermal, mechanical and chemical stability. There are lot of reports on preparation of graphene oxide (GO) and reduced graphene oxide (R-GO) by hummers method, mechanical exfoliation, microwave synthesis, colloidal suspension and chemical vapor deposition (CVD) method. The graphene nanosheets have been widely used in various fields, which include thin film devices, nanoelectronics, flexible transparent electronic devices, bio-sensors and high energy storage in supercapacitor electrode applications.

Recent development includes the doping in carbon nanostructures such as Carbon nanotubes (CNTs) and graphene nanosheet (GNS) with heteroatoms such as Nitrogen (N), Boron (B), Phosphorus (P), Sulfur (S) and Flourine (F). The doping in carbon materials further increases



the semiconducting and conducting properties of these materials for energy storage and conversion application. The doping of heteroatoms in graphene and CNTs will enhance the properties in two possible ways 1) Doping in covalent bonding in carbon lattice into heteroatoms like gas molecules due to thermal effect 2) The chemically substitutional doping into hexagonal carbon lattice. The chemical functionalization and heteroatom doping of graphene sheets helps to tune the band gap, thereby enhancing the electrochemical properties of graphene sheets towards electrochemical energy storage applications. Chemically modified or functionalized doping of graphene is either p-type or n-type, depending on the electronic structure and the type of guest atoms. Heteroatoms can be covalently bonded to graphene lattice. Understanding the characteristics of doped graphene help us to study the electronic band structure due to heteroatom bonding with different sites and the chemical properties using electrochemical analysis for energy storage and conversions applications.

## **ELECTROCHEMICAL SYNTHESIS OF NITROGEN DOPED FEW LAYER GRAPHENE FOR SUPERCAPACITOR APPLICATIONS**

Lately, graphene readiness is centered on two dimensional carbon nanostructures materials and it comprises of mono layer carbon iotas orchestrated thickly and firmly pressed honeycomb like hexagonal construction. They have centered different applications in hardware, electrochemical energy stockpiling and energy discussion gadgets because of their great electrical properties, high warm steadiness, fantastic mechanical strength and high surface region. There are well known strategies for the synthesis of single just as not many layer graphene by synthetic peeling technique, substance fume testimony strategy and microwave technique. Graphene has the most encouraging applications in supercapacitors anode materials for high energy stockpiling gadgets, which was first investigated by Ruoff and colleagues. The supercapacitor, otherwise called ultracapacitors, which are arranged into three principle classes: (I) electrical twofold layer capacitor (EDLC), (ii) pseudocapacitor and (iii) half and half capacitors. Supercapacitors have high power thickness with scaled down size and light weight than the Lithium particle batteries.

These days, heteroatom doped nanostructured materials have been generally utilized for the supercapacitor applications because of high conductivity. Heteroatoms like boron, nitrogen, sulfur, fluorine and phosphorus doped carbon nano materials change their unique design,

primary deformities and their gadgets properties. The imperfections/underlying changes could be positive to produce novel new nanostructured materials for energy stockpiling gadget applications.

Nitrogen doped single and few layer graphene was integrated by various methodologies by Chemical fume affidavit, aqueous strategy, Solvothermal Method and nitrogen plasma treatment.

## CONCLUSION

The postulation has extended the facile and novel method of synthesis, portrayal and the applications of the N - doped coiled CNTs and N, B, P doped graphene. This proposition is separated into six parts and the results are summed up section astute with the critical conclusion as follows. The presents the successfully blend of nitrogen doped open finished helical formed CNTs onto Ni slender film covered on S.S foil substrate by thermal T-CVD strategy. The blended carbon nanotubes were described and utilized as electrode material for electrochemical examinations. Coiled N-DWCNTs was set up with acetylene ( $C_2H_2$ ) and simultaneously flow of was alkali ( $NH_3$ ) (99 %) gas. SEM and TEM picture of unadulterated CNTs explain the tip development system. The cross part of unadulterated CNT shows the backwoods development with few aligned CNTs. • The FT-IR analysis was done for the samples and the functional gatherings in N DWCNTs and Pure CNTs were recognized. It also affirms the presence of nitrogen functional gatherings in N-CNTS. The CV plots show nearly rectangular shapes with oxidation and decrease tops for all the output rates. The redox responses are because of the nitrogen doping in the CNTs and furthermore affirms the faradic redox response with pseudocapacitive impacts because of the presence of Nitrogen. Perhaps the best test of the present remote, portable society is to give exceptionally productive, minimal effort, and harmless to the ecosystem energy stockpiling gadget for expanding different applications.

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